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INFORMATION REPORT

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COUNTRY USSR

SUBJECT

Soviet Research on Synthesis and Insecticidal

Properties of Some Mixed Esters of Dithiophosphoric Acid

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I have recently come across an article entitled "Synthesis of Some Mixed Esters of Dithiophosphoric Acid" by N N Mel'nikov and K D Shvetsova-Shilovakaya, appearing in Doklady Akad Nauk, SSSR, 86,543-6 (1952). This article lists the properties of an entire series of phosphorous compounds similar to Parathion, an insecticide toxic to humans and a cholinesterase inhibitor. The phosphorous compounds listed in this article are also most likely to be toxic to humans and cholinesterase inhibitors. Of course, no data are given in the article on this phase, and I would be quite surprised if data had been given. Indeed, there is a fair assumption which can be made that probably some data on these compounds would have been given in the article, provided the compounds were not toxic to

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In the insecticide business it is impossible to tell whether compounds possess toxicky without actually experimenting with them on mice, rabbits, etc. This is done on a basis of body weight in which the results are analyzed and applied to human weights.

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This article is the latest in a series of articles having to do with organophesphorous compounds similar to Parathion, beginning with an article published in 1950 by N N Mel'nikov, one of the coauthors of the nurrent article, and two of his colleagues, Ya A Mandel'daum and P V Popcy, entitled "Synthesis and Insecticidal Properties of Some Esters of Phosphorit Add" The 1950 article is the first acknowledgment in the Soviet press of compounds related to Parathion. The 1950 article is the first acknowledgment

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Shvetsova-Shilovskaya as follows:

the 1952 article by Mel'nikov and

It was shown that (RO)2PS2H add to unsatd compounds (PhCH:CH2, CH2:CHCO2Me, CH2:CHCN, CH2:CHOAc, CH2:CHCHO, esters of maleic acid, etc) yielding the mixed esters of dithiophosphoric acid; the reaction occurs in the cold in the presence of small amounts of solid KOH. For best results 1 mole of unsatd compound was treated with (RO)2PS2H at such rate as to keep temp at 40-50; after standing at room temp until reaction ceased (titration of free acid with O.1 N NaOH) the products were distd. The following are reported: (MeO)2PE2CH(CO2Et)CH2CO2Et, 50%, b_{3.5}160-70°, n²⁰_D1.4960, d₄²⁰1.2076; P,P-di-Et analog, 59%, b₃157-62°, 1.4910, 1.1742; P,P-d1-Pr enalog, 27%, b_{0.1}145°, 1.4880, 1.1706; P,P d1-1so-Pr analog 39%, b₄151°, 1.5440, 1.0702; P,P di-Bu analog, 40%, b_{0.025}125-8°, 1.4861, 1.1078; P,P di-iso-Bu analog, 15%, b_{0.04}117-30°, 1.4555, 1.0642; (Eto) PS2CH2CH2CH, 33%, b_{3.5}137-42°, 1.5195, 1.1704; P,P d1-Pr analog, 30%, b_{0.05}116-20°, 1.5068, 1.0505; P,P-d1-Bu analog, 27.5%, b_{0.03}121-3°, 1.5050, 1.0816; P,P d1-iso-Bu analog, 44%, b_{0.05}122-3°, 1.5010, 1.0986; (MeO)₂PS₂CH₂CH₂O₂CMe, 38%, b₂100-3°, 1.5252, 1.1558; P,P d1-Et analog, 71%, b_{1.5}115-17°, 1.4948, 1.1517; P,P d1-Pr analog, 46%, b_{0.05}80-1°, 1.5075, 1.0984; P,P d1-1so-Pr analog, 50%, b_{0.1}72-4°, 1.4335, 1.846 (1.0846-?); P,P di-180-Bu analog, 43%, b_{0.075}104°, 1.4858, 1.0926; P,P di-Bu analog, 25%, b_{0.4}109°, 1.4915, 1.0915; (MeO)₂PS₂CHMePh (presumably)50%, b_{0.45}-128-32°, 1.5705, 1.2108; P,P d1-Et analog, 35%, b_{0.15}135-7°, 1.5498, 1.1444; P,P di-Pr analog, 38.5%, b 129-31°, 1.5381, 1.0990; P,P di-iso-Pr analog, 38.5%, b_{0.2}121-4°, 1.5365, 1.0966; P,P di-Bu analog, 13%, b_{0.025}137-40°, 1.5320, 1.0890; P,P di-1so-Bu analog, 17.5%, b_{0.04}117-22°, 1.5301, 1.0860; (MeO)₂PS₂-CH₂CH₂CO₂Me, 67%, b₃145°, 1.5160, 1.2625; P,P di-Et analog, 83%, b 1.5167°, 1.5050, 1.1911; P,P d1-iso-Bu analog, 58%, b 124-9°, 1.4915, 1.1162; (MeO)2-PS₂CH₂CHMeCO₂Me (presumably), 384, b₂133-5°, 1.5100, 1.2330; P,P di-Et analog, 59%, b₅154.5°, ± 4995, 1.1577; P,P d1-180-Pr analog, 47%, b_{0.02}84-50, 1.4935, 1 1203; P,P di-icc-Du analog 56%, b_{0.04}115-16°, 1.4915, 1.1138; P,P di-Bu analog, 38%, b_{0.025}110°, 1.4918, 1.1132; adduct of (isc-Pro)₂PS₂H and CH₂:CHCHO, 37%, b0.15740, 1.5040, 1.1348. It is stated that (Bu0)2PS2CH2CH202CMe lost CO2 during disin and the analytical data are given for the "decarboxylated ester".

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It is not clear whether or not the constants cited are for the decarboxylated material.

